















- The dewpoint temperature is the temperature to which a given air parcel must be cooled at constant pressure and water vapor content in order for saturation to occur.
- Substitute e for e<sub>s</sub> and T<sub>d</sub> for T in the Clausius-Clapeyron equation:

$$e = e_{s_0} \exp\left[\frac{L}{R_v}\left(\frac{1}{T_0} - \frac{1}{T_d}\right)\right]$$











## Requirements for a successful psychrometric measurement

- Two well-matched thermometers
- Adequate ventilation (> 3 m s<sup>-1</sup>)
- Radiation shield
- Distilled water to moisten the wick
- Dissolved salts affect the evaporation rate
- Clean wick
  - Special psychrometer wick with no hydrophobic chemicals (not cotton)





## Equilibrium sorption of water vapor

- Many hygrometers use the process of sorption to measure water vapor
  - Ab<u>sorption</u>: uptake of water into the bulk of the substance
  - Ad<u>sorption</u>: surface retention of water molecules
    Mass of water is proportional to relative humidity
- Consequences of sorption processes:
- Material expands/contracts
- Resistance or capacitance of material changes

























